1. The synthesization steps of DADMAC and PolyDADMAC

In general, PolyDADMAC is manufactured in two consecutive steps: synthesization of monomer DADMAC and then polymerization of this monomer [25]. Meanwhile, the monomer DADMAC is synthesized by two chemical reactions of dimethylamine and allyl chloride. The dimethylamine is firstly alkylated by allyl chloride in an aqueous alkaline solution, followed by a quaternization step where the dimethylamine reacts with an extra amount of allyl chloride in an organic medium (Figure S1: a, b). PolyDADMAC has a unique chemical structure where it owns a long backbone of cyclic units resulting from the polymerization of DADMAC (figure S1: c).

(a)
$$CH_{3} \longrightarrow NH + CICH_{2} - CH = CH_{2} \xrightarrow{+ NaOH} CH_{3} \longrightarrow N - CH_{2} - CH = CH_{2}$$

$$CH_{3} \longrightarrow N - CH_{2} - CH = CH_{2} + CICH_{2} - CH = CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{2} - CH = CH_{2}$$

$$CH_{3} \longrightarrow N - CH_{2} - CH = CH_{2} + CICH_{2} - CH = CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{2} - CH = CH_{3}$$

$$CH_{3} \longrightarrow CH_{2} - CH = CH_{2} \longrightarrow CH_{2} - CH - CH_{2} - CH = CH_{3}$$

$$CH_{3} \longrightarrow CH_{2} - CH = CH_{2} \longrightarrow CH_{2} - CH_{2} - CH_{2} - CH_{2} \longrightarrow CH_{2} - CH_{2} - CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{2} - CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{2} - CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{2} - CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{3} \longrightarrow CH_{2} \longrightarrow CH_{3} \longrightarrow CH_{3$$

Figure S1: (a) The alkylation of dimethylamine with allylchloride [40], (b) Formation of DADMAC by quaternization of dimethylamine with extra allylchloride [40], and (c) Formation of PolyDADMAC by the polymerization of DADMAC [25]

2. Preparation of Draw Solutions

$$C_i X V_i = C_f X V_f \tag{S1}$$

where $C_i(g/mL)$ is the stock solution concentration. V_i (mL) is the required stock solution volume. C_f (g/mL) is the final solution concentration. And V_f (mL) is the final solution volume after dilution.

Table S1: Required volumes to prepare 2000 mL of draw solutions with various concentrations

Final	Required volume of	Required volume of
concentration	stock DADMAC	stock PolyDADMAC
g/mL	mL	mL
0.035	103.55	183.49
0.085	251.48	445.61
0.120	355.03	629.10
0.155	458.58	812.58